THIRTEENTH ANNUAL CONFERENCE

YUCOMAT 2011

Hunguest Hotel Sun Resort Herceg Novi, Montenegro, September 5-9, 2011 http://www.mrs-serbia.org.rs

Programme and The Book of Abstracts

Organised by:

Materials Research Society of Serbia

under the auspices of
Federation of European Material Societies
and
Materials Research Society

Title: THE THIRTEENTH ANNUAL CONFERENCE

YUCOMAT 2011

Programme and The Book of Abstracts

Publisher: Materials Research Society of Serbia

Knez Mihailova 35/IV, 11000 Belgrade, Serbia Phone: +381 11 2185-437; Fax: +381 11 2185-263

http://www.mrs-serbia.org.rs

Editor: Prof. Dr. Dragan P. Uskoković

Technical editor: Aleksandra Stojičić

Cover page: Aleksandra Stojičić and Milica Ševkušić

Copyright © 2011 Materials Research Society of Serbia

Acknowledgment:



Printed in: Biro Konto

Sutorina bb, Igalo – Herceg Novi, Montenegro

Phones: +382-31-670123, 670025, E-mail: bkonto@t-com.me Circulation: 250 copies. The end of printing: August 2011

THIRTEENTH ANNUAL CONFERENCE YUCOMAT 2011 Herceg Novi, September 5-9, 2011

O.S.A.1.

SONOELECTROCHEMISTRY – A VERSATILE TOOL FOR THE PREPARATION OF NANOMATERIALS

Chr. Argirusis^{1,2}, S. Martens², P. Sakkas¹, G. Sourkouni^{2,3}, O. Schneider^{2,4}

¹National Technical University of Athens; School of Chemical Engineering, Zografou, Greece,

²Clausthal University of Technology, Institute of Metallurgy, Clausthal-Zellerfeld, Germany

³Clausthal University of Technology, Energy Research Centre Lower Saxony, Goslar, Germany

⁴Technische Universität München, Department of Physics, Chair for Energy Conversion and

Storage, Garching, Germany

Sonoelectrochemistry, the application of ultrasound in electrochemistry has become increasingly popular in the past decade, especially in the field of materials science because of its potential in the production of nanomaterials, metals, alloys and composites. The major effects of ultrasound in electrochemical reactions are the enhanced mass transport due to acoustic streaming, influence of cavitation on reaction mechanism and surface cleaning and erosion. Electrochemical experiments can be performed by positioning the horn face to face to the working electrode with a distance between 1-80 mm, or by using the ultrasonic horn generating the ultrasound itself as the cathode, which is especially suitable for the production of nanopowders. Both approaches are discussed along with the possibility to use an Electrochemical Quartz Crystal Microbalance to monitor the process.

O.S.A.2.

SYNTHESES AND CHARACTERIZATIONS OF LiFePO₄ POWDERS

<u>D. Jugović</u>¹, M. Kuzmanović¹, M. Mitrić², N. Cvjetićanin³, D. Uskoković¹

Institute of Technical Sciences of SASA, Belgrade, Serbia,

²The Vinča Institute of Nuclear Sciences, University of Belgrade, Serbia,

³Faculty of Physical Chemistry, University of Belgrade, Serbia

The olivine type compositions LiMPO₄ (M = Fe, Mn, Co) are among the most attractive materials for the positive electrode of lithium-ion battery. The benefits of using LiFePO₄ are excellent cycle life, high structural stability, low cost and environmental friendliness. Here will be presented our efforts to obtain LiFePO₄ powders through several synthesis methods. The differences in phase purity, microstructure, morphology, and electrochemical performances of synthesized powders were investigated by using several techniques such as X-ray powder diffraction, scanning electron microscopy, galvanostatic cycling, Mossbauer spectroscopy, and thermogravimetry. It was found that carbon coated, nanostructured particles with minimal structural imperfections showed the best electrochemical performances.