

## Serbian Ceramic Society Conference ADVANCED CERAMICS AND APPLICATION X New Frontiers in Multifunctional Material Science and Processing

### Serbian Ceramic Society Institute of Technical Sciences of SASA Institute for Testing of Materials Institute of Chemistry Technology and Metallurgy Institute for Technology of Nuclear and Other Raw Mineral Materials

# **PROGRAM AND THE BOOK OF ABSTRACTS**

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material. Here, for the first time two combinations of rare earth dopant ions,  $Yb^{3+}/Ho^{3+}$  and  $Yb^{3+}/Tm^{3+}$ , with different mutual ratios were chosen as pairs for inducing up-conversion. Solgel assisted combustion synthesis, which comprises citric acid as chelator and glycine as fuel, was used to obtain powdered samples that are subsequently thermally treated for 3.5 h at 1100°C. X-ray powder diffraction analysis (XRPD) was performed to determine crystal structure. Morphology characteristics were observed by scanning and transmission electron microscopy (SEM/TEM). Photoluminescent up-converting properties were measured in function of laser power (976 nm) in order to define optimal doping concentration and up-conversion mechanism.

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#### **Related effects of Ca<sub>1-x</sub>Gd<sub>x</sub>MnO<sub>3</sub> (x = 0.05, 0.1, 0.15, 0.2) compound**

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We have used theoretical and experimental methods to investigate the octahedral tilting and related effects of  $Ca_{1-x}Gd_xMnO_3$  (x=0.05, 0.1, 0.15, 0.2) compound. Both methods have shown that orthorhombic-perovskite structure (space group *Pnma*) is the most stable form and according to Glazer's classification belongs to  $a^{-}b^{+}a^{-}$  tilt system. Our bond valence calculations (BVC) have shown ten additional perovskite-related modifications of the equilibrium  $Ca_{1-x}Gd_xMnO_3$  structure, and their stability has been investigated as function of Gd doping. We have further studied the influence of gadolinium amount on Mn-O bond angles and distances, tilting of  $MnO_6$  octahedra around all three axes and deformation due to the presence of Jahn-Teller distortion around  $Mn^{3+}$  cation, and calculated the amount of  $Mn^{3+}$  in the system. The infrared reflection spectra of  $Ca_{1-x}Gd_xMnO_3$  samples confirmed XRD results that  $Ca_{1-x}Gd_xMnO_3$  nanopowders are of *Pnma*-1 structure and that the tilting of octahedra are increased with Gd doping. The EPR spectra are in accordance with the assumption that EPR linewidth is Mn-O-Mn angle dependent. The studied samples showed that small octahedra tilting in these samples brought only a small change of the EPR linewidth.